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(54) Title: ACID-LABILE ISOTOPE-CODED EXTRACTANT (ALICE) AND ITS USE IN QUANTITATIVE MASS SPECTROMETRIC ANALYSIS OF PROTEIN MIXTURES

(57) Abstract: The method of the invention provides novel compounds, termed acid-labile isotope-coded extractants (ALICE), for quantitative mass spectrometric analysis of protein mixtures. The compounds contain a thiol-reactive group that is used to capture cysteine-containing peptides from all peptide mixtures, an acid-labile linker, and a non-biological polymer. One of the two acid-labile linkers is isotopically labeled and therefore enables the direct quantitation of peptides/proteins through mass spectrometric analysis. Because no functional proteins are required to capture peptides, a higher percentage of organic solvent can be used to solubilize the peptides, particularly hydrophobic peptides, through the binding, washing and eluting steps, thus permitting much better recovery of peptides. Moreover, since the peptides are covalently linked to the non-biological polymer (ALICE), more stringent washing is allowed in order to completely remove non-specifically bound species. Finally, peptides captured by ALICE are readily eluted from the polymer support under mild acid condition with high yield and permit the direct down stream mass spectrometric analysis without any further sample manipulation. In combination with our novel dual column two dimensional liquid chromatography-mass spectrometry (2D-LC-MS/MS) design, the ALICE procedure proves to a general approach for quantitative mass spectrometric analysis of protein mixtures with better dynamic range and sensitivity.

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# ACID-LABILE ISOTOPE-CODED EXTRACTANT (ALICE) AND ITS USE IN QUANTITATIVE MASS SPECTROMETRIC ANALYSIS OF PROTEIN MIXTURES

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## BACKGROUND OF THE INVENTION

The present invention relates to the field of high-throughput quantitative protein analysis and, more specifically, to novel reagents for use in such analysis.

Most approaches to quantitative protein analysis are accomplished by  
10 combining protein separation, most commonly by high-resolution two-dimensional polyacrylamide gel electrophoresis (2D-PAGE), with mass spectrometry (MS)-based sequence or tandem mass spectrometry (MS/MS)-based sequence identification of selected, separated protein species.

S. P. Gygi, et al., *Nature Biotech*, 17:994-999 (October 1999) describes an  
15 approach to quantitative protein analysis based on a class of reagents termed isotope-coded affinity tags (ICAT), which consist of three functional elements: a specific chemical reactivity, an isotopically coded linker, and an affinity tag. The reagents described by Gygi utilize biotin as the affinity tag and rely upon biotin-avidin affinity binding to isolate the cysteine-containing peptides from the complex peptide mixture.

20 Although the ICAT approach has many advantages over the traditional 2D-PAGE/MS approaches, it does possess some intrinsic limitations. For example, ICAT adds a relatively large chemical moiety onto the cysteine-containing peptides and this functionality is very labile under collision induced dissociation (CID) condition and thus complicates the downstream data analysis. Non-specific binding is also a concern  
25 since the enrichment relies on non-covalent affinity binding between a protein (avidin) and the biotinylated peptides. Finally, the captured peptides are not readily eluted from the avidin beads with high recovery using MS-compatible conditions. Thus, there is a need in the art for additional reagents and methods for improving performance in quantitative mass spectrometric analysis of protein mixtures.

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## SUMMARY OF THE INVENTION

The invention provides polymer-based compounds useful for quantitative analysis of mixtures containing proteins. Advantageously, the compounds of the invention bind covalently with the peptides which they are used to tag, permitting the tagged peptides to be subjected to more rigorous washing techniques. Thus, the tagged peptides are more readily purified, without nonspecifically bound species. This results in lower background on MS spectra and thus provides an increase of dynamic range and sensitivity in quantitation and identification of the proteins.

In one aspect, the invention provides a method for the quantitative analysis of mixtures containing proteins. The method involves (a) reducing the disulfide bonds in the proteins of a sample to provide free thiol groups in cysteine-containing proteins; (b) blocking free thiols on the reduced proteins with a blocking reagent; (c) digesting the proteins in the sample using an enzyme such as trypsin; (d) reducing the peptides following the digestion step; (e) reacting cysteine-containing peptides with a reagent, wherein the reagent comprises a thiol-specific reactive group covalently bound to a polymer tag via a linker, wherein the linker can be differentially labeled with stable isotopes (optionally prior to or following any of the reduction steps); (f) washing the polymer-bound peptides to remove non-covalently bound compounds; (g) eluting the cysteine-containing peptides; and (h) subjecting the retrieved peptides to quantitative mass spectrometry (MS) analysis. In one embodiment, the method further involves performing steps (a) to (d) on a second sample; reacting cysteine-containing peptides in the second sample with a stable isotope-labeled form of the reagent, wherein in reacting step (e), the reagent used is a non-isotope labeled form of the reagent; mixing the peptides of the reacted sample following step (e) and the reacted second sample; and performing steps (g) and (h) on the peptides in the mixture.

In another aspect, the invention provides a compound useful for capturing cysteine-containing peptides. This compound is composed of a thiol-specific reactive group attached to a non-biological polymer via a linker. In one desirable embodiment, the reagent has the formula: A1 - Linker - A2 - polymer, wherein A1 is a thiol-reactive group and A2 is an acid labile group to which the polymer is attached.

In yet another aspect, the invention provides a reagent kit for the mass spectral analysis of proteins that comprises a compound of the invention.

Other aspects and advantages of the present invention are described further in the following detailed description of the preferred embodiments thereof.

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### BRIEF DESCRIPTION OF THE DRAWINGS

Fig. 1A and Fig. 1B provides a schematic of the automated 2D-LC/MS System of the invention.

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### DETAILED DESCRIPTION OF THE INVENTION

The present invention provides a novel approach for the quantitative analysis of proteins using acid-labile isotope coded extractants (ALICE) which are useful for capturing cysteine-containing peptides. The advantage of this approach over the prior art, is that it replaces biotin-avidin affinity binding with acid-labile covalent binding to retrieve cysteine-containing peptides from the mixture. Since the binding is covalent, more stringent detergents or organic solvents can be used during the procedure to keep hydrophobic proteins and peptides in the solution and thus maximize the overall peptide recovery. Furthermore, the compounds and method of the invention avoid nonspecific peptide-protein binding. Removal of all detectable non-covalently bound species during the washing step(s) is also accomplished. Thus, the final cysteine-containing peptide solution is much less contaminated, resulting in higher sensitivity and dynamic range of MS analysis. Lastly, since the ALICE label is small in size and does not undergo fragmentation during MS/MS analysis, it does not interfere with the downstream MS analysis and database searching.

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In one embodiment, the present invention provides a compound of the formula: A1 - Linker - A2 - polymer, wherein A1 is a thiol-reactive group and A2 is an acid labile group to which the polymer is attached. Alternatively the acid labile group may be absent and the polymer may be attached directly to the linker.

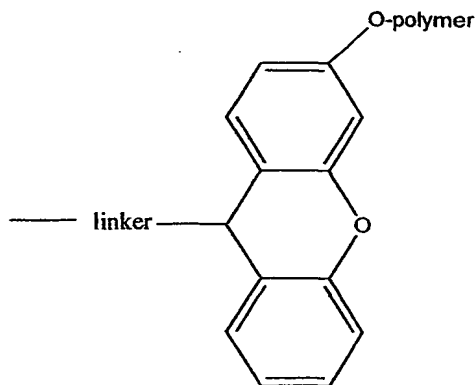
Most preferably, the polymer is a non-biological polymer. As used herein a non-biological polymer includes inorganic polymers and organic polymers which

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form a covalent bond with the acid-labile group, where present, or the linker.

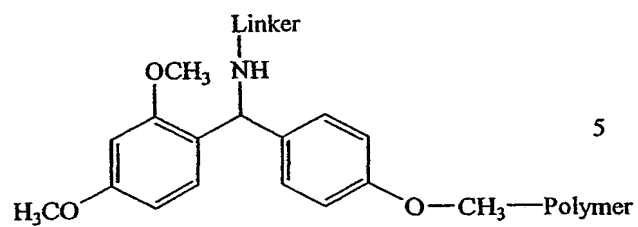
Suitably, an organic polymer selected does not interfere with the process steps in the method of the invention, e.g., is stable under basic conditions and in the presence of the detergents and/or organic solvents required to maintain the mixture in solution. In one suitable embodiment, the polymer used in the invention is a solid substrate composed of a homopolymer or a heteropolymer containing polystyrene, polyethylene, polyacrylamide, polyacrylein, polyethylene glycol, or the like. Suitable polymers and solid substrates, e.g., resins, beads or the like, are available from a variety of commercial sources including Sigma-Aldrich, NovaBiochem, and Beckman-Coulter, or may be synthesized using known techniques. An example of one suitable synthesis technique is provided in Example 1 below. However, the invention is not so limited.

In one embodiment, the polymer is covalently bound to the linker via an acid-labile group that provides the compound of the invention with the ability to be readily eluted using an acidic reagent. In one preferred embodiment, the acid-labile group bound to the polymer has the following structure:

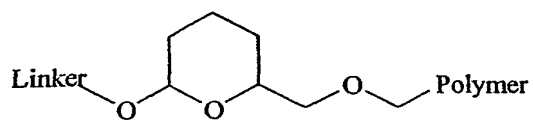


in which the linker is  $\text{-CONH-}$ ,  $\text{-COO-}$ , or another amide or ester. However, other structures can be readily synthesized to contain other suitable groups that provide similar qualities to the compound in terms of stability and accessibility to acid elution. Examples of suitable acid-labile groups include:

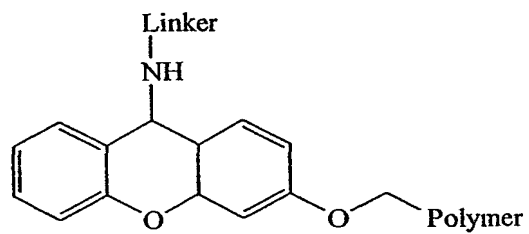
Rink Amide Linker:



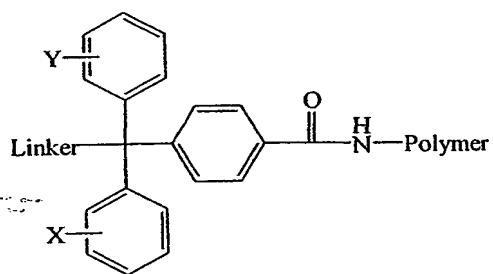
DHP Linker:



**Siber Linker:**

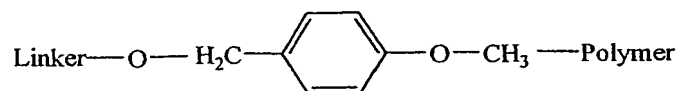


**Trityl Linker:**



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Wang Linker:



In certain embodiments, this function may be provided by the linker, and the acid labile group may be absent.

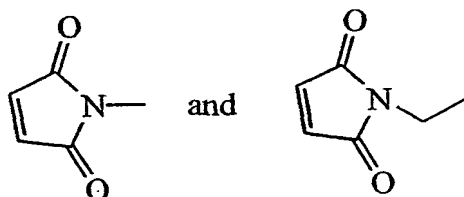
The linker is any structure that may be differentially labeled with stable isotopes for use in MS techniques. In one embodiment, the linker contain from 1 to 100 atoms in length, about 3 to about 50 atoms in length, or about 5 to about 15 atoms in length, which are composed of carbon, and optionally, one or two atoms selected from O, S, NH, NR, NR', CO, C(O)O, C(O)S, S-S, SO<sub>2</sub>, C(O)-NR', CS-NR', or Si-O. Optionally, one or more of the C atoms may be substituted with a small alkyl (C<sub>1</sub>-C<sub>6</sub>), alkenyl, alkoxy, aryl, or diaryl groups. For example, the linker may be an alkyl, alkenyl, or alkynyl group, optionally substituted as described above. In another example, the linker may itself contain one or more O, S, NH, NR, NR', CO, C(O)O, C(O)S, S-S, SO<sub>2</sub>, C(O)-NR', CS-NR', Si-O groups bound to one or more C atoms, which may be optionally substituted.

In one embodiment, the linker is a structure (e.g., an alkyl group) which contains a substitution of about four to about twelve atoms with a stable isotope. However, in certain embodiments, it is desirable for the linker to contain substitutions of at least six atoms with a stable isotope. For example, for peptides at the higher end of the molecular weight range at which MS is useful (e.g., about 2000 Da to 3500 Da) it may be desirable for the linker to contain eight, ten, twelve or more substitutions, in order to achieve the differential analysis required; whereas peptides at the lower end of the molecular weight range for MS (e.g., about 500 to 2000 Da) may require only four to six substitutions. For the selected number of substitutions, any one or more of the hydrogen, nitrogen, oxygen, carbon, or sulfur atoms in the linker may be replaced with their isotopically stable isotopes: <sup>2</sup>H, <sup>13</sup>C, <sup>15</sup>N, <sup>17</sup>O, <sup>18</sup>O, or <sup>34</sup>S.

Thus, the linker group has a structure that accommodates the number of isotope substitutions desired. The selection of this structure is not a limitation of the present invention. One or more of the atoms in the linker can be substituted with a stable isotope to generate one or more substantially chemically identical, but isotopically distinguishable compounds. Additionally or alternatively, the linker also optionally provides desired acid labile properties to the compound.

The compound of the invention further contains a functional group that is reactive, preferably specifically, with cysteine residues. Desirably, the reactive group is selected from the group consisting of either maleimide (see below)

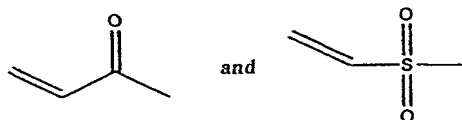
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or  $\alpha$ -haloacetyl groups such as  $X-CH_2CO-$ . Most suitably, the X is selected from halogens such as iodine, bromine, and chlorine to form iodoacetyl, bromoacetyl, or chloroacetyl functionalities.

In another alternative, the thiol-reactive group may be selected from other  $\alpha$ -,  $\beta$ -conjugated double bond structures, such as

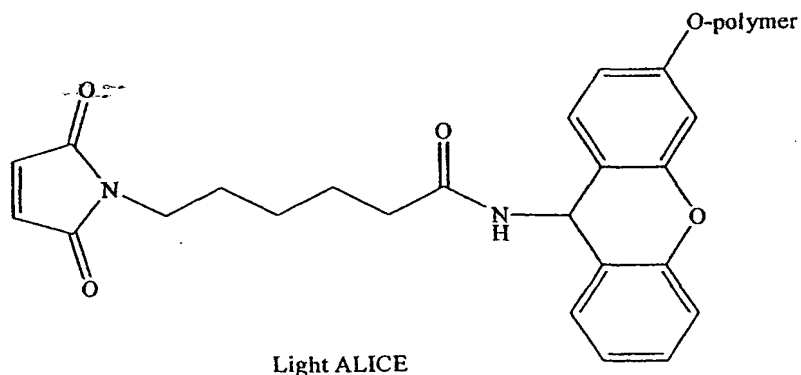
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and the like. Still other reactive group can readily be synthesized to contain other thiol-specific reactive groups for use in binding cysteine-containing peptides.

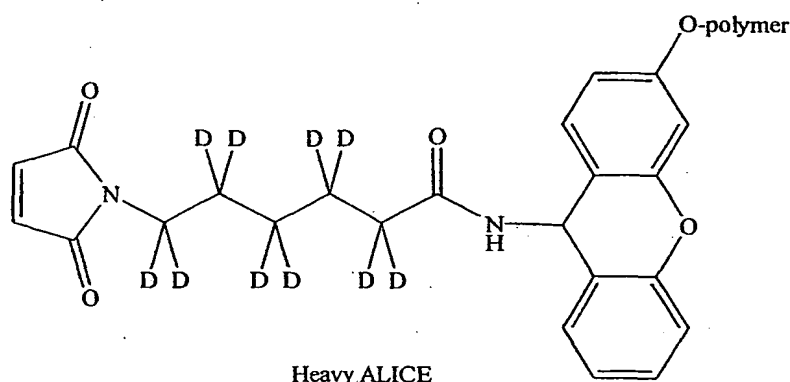
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In one preferred embodiment, a compound of the invention has the formula:



In one desirable embodiment, this compound is isotopically modified as follows.





However, the invention is not so limited. One of skill in the art can readily provide  
 5 light ALICE with other stable isotopes. Further, one of skill in the art can readily  
 produce other suitable compounds in view of the guidance provided herein.

#### METHOD OF USING THE COMPOUNDS OF THE INVENTION

The compounds of the invention are particularly useful in mass spectrometric  
 10 methods for quantitation and identification of one or more proteins in a mixture. The  
 peptides analyzed by the method of the invention are most preferably about 500  
 Daltons (Da) to about 3500 Da in size, but may be larger. Suitably, these peptides are  
 formed upon enzymatic digestion of proteins in a complex mixture. The protein  
 mixture may be a sample from a cell or tissue culture, or biological fluids, cells or  
 15 tissues. Samples from a culture include cell homogenates and cell fractions.  
 Biological fluids include urine, blood (including, e.g., whole blood, plasma and sera),  
 cerebrospinal fluid, tears, feces, saliva, and lavage fluids. The mixtures may include  
 proteins, lipids, carbohydrates, and nucleic acids. The methods of the invention  
 employ MS and (MS)<sup>n</sup> methods. Currently, matrix assisted laser desorption ionization  
 20 MS (MALDI/MS) and electrospray ionization MS (ESI/MS) methods are preferred.  
 However, a variety of other MS and (MS)<sup>n</sup> techniques may be selected.

In one embodiment, the invention provides a method for quantitative analysis of a proteome using the compound of the invention. Typically, a sample is obtained from a source, as defined above. The sample may be compared to a reference protein mixture, which is obtained as a sample from the same source or may be obtained from another source. Where a sample protein mixture is to be compared to a second sample or a reference protein mixture, these mixtures are processed separately, applying identical reaction conditions, with the exception that one sample will be reacted with the compound containing heavy stable isotopes. Where samples are not to be compared, separate processing to the point of reaction with the compound(s) of the invention is not necessary, but is permitted.

Typically, the protein sample is solubilized in a suitable buffer that may contain an organic solvent. Throughout the entire procedure except the final peptide elution step, the pH of the mixture is maintained under basic conditions. Most suitably, the pH is maintained between 6.5 and 9, more preferably about 7.5 to 8.5, and most preferably about 7.2 to 7.5.

The disulfide bonds of the proteins in the sample(s) or reference mixtures are reduced to free SH groups. Optionally, this step may be combined with solubilization of the protein or protein mixture, referred to above. Suitable reducing agents include tri-n-butylphosphine (TBP), 2-mercaptoethanol, dithiothreitol, and tris-( $\beta$ -carboxyethyl) phosphine. However, other suitable reducing agents may be substituted. In one embodiment, disulfide bonds in 2 mg of a protein are denatured using 8M urea, 200 mM ammonium bicarbonate, 20 mM  $\text{CaCl}_2$ , 5  $\mu\text{mole}$  TBP, which has been pre-dissolved in 20  $\mu\text{L}$  of acetonitrile (ACN) and incubated for one hour at about 37°C. In another embodiment, a protein may be incubated in 50 mM Tris buffer, 6 M guanidine-HCl, 5 mM TBP at pH 8.5 for 1 hour at 37°C. However, other concentrations of these components and/or other reducing agents, buffered to a pH in the basic range may be selected and incubated for varying lengths of times.

Free thiols (SH) are blocked using a suitable blocking reagent, e.g., methyl methane thiosulfonate (MMTS), which functions under the basic conditions provided and does not interfere with the performance of the following steps. Although MMTS

is preferred, other suitable blocking reagents, including, without limitation, o-methylisourea, may be selected by one of skill in the art.

The proteins in the samples are enzymatically digested. A suitable protease for use in this method may be readily selected from among proteases that are compatible with the basic conditions and the procedure. Under certain circumstances, it may be necessary to dilute the sample mixture until any denaturing solubilizing agents in the sample are diluted to a point at which they are compatible with the activity of the protease or proteases used. In one embodiment, the protease is trypsin. In another embodiment, the protease is the endoproteinase Lys-C (commercially available, e.g., from Promega, Roche Molecular Biochemical). In still another example, a mixture of proteases that have similar activity levels at basic pH is used. Such proteases may include aminopeptidases, carboxypeptidases, among others. Alternatively, the protein mixture is subjected to more than one digestion step. For example, the protein mixture may be subjected to digestion with Lys-C, followed by digestion with trypsin. Multiple digestions are particularly desirable where the mixture is a complex mixture. One of skill in the art can readily determine whether a single digestion step, or multiple steps, are required. In yet another alternative, protein digestion may be omitted where the sample contains peptides, polypeptides or small proteins (e.g., about 500 to 5000 Da).

Suitably, the peptides are again reduced prior to being reacted with the compounds of the invention to remove the blocking reagents. The reduction step is performed using the reagents described above. In one suitable embodiment, the mixture is reduced by incubation with 5  $\mu$ mole of TBP at 37°C for one hour. However, other suitable concentrations, reagents, incubation temperatures and times may be readily substituted.

A selected compound of the invention and a corresponding isotopically heavy compound are reacted with the samples. Typically, the reference sample is labeled with the isotopically heavy compound and the experimental sample(s) are labeled with the isotopically light form of the compound. However, the labeling may be reversed.

Optionally, this labeling reaction may be performed at any stage of the method, e.g., prior to any of the reduction steps.

After completion of the tagging reaction, defined aliquots of the samples labeled with isotopically different compounds (e.g., corresponding light and heavy compounds) are combined and all the subsequent steps are performed on the pooled samples. Preferably, equal amounts of each sample are pooled.

The pooled samples are washed in order to remove any non-covalently bound species. The use of the compounds of the invention permits the use of harsher washing steps than prior art reagents can withstand. For example, one suitable method utilizes 5 X 1 mL of 50% acetonitrile (ACN), 5 X 1 mL of 30% ACN, 5 X 1 mL of 90% ACN, 5 X 1 mL (non-diluted) ACN, and 10 X 5 mL dichloromethane. However, the concentration of ACN may be varied. Alternatively, other suitable solvents may be substituted. Examples of suitable solvents include organic solvents with polarity properties similar to acetonitrile or dichloromethane. Yet another suitable method utilizes high concentrations of organic solvents, which effectively removes any residual detergents or surfactants.

The tagged peptides are selectively retrieved by acid elution, which breaks the bond between the linker or acid labile group and the polymer to which it is covalently bond allowing the peptides tagged with the light or heavy compounds of the invention to be eluted. For example, the last washing may be eluted using 1% to 5% trifluoroacetic acid (TFA) in dichloromethane ( $\text{CH}_2\text{Cl}_2$ ). Using the method of the invention, peptide recovery is estimated at above 75%. Suitably, recovery may be even higher, e.g., above 80%, 85%, and 90%, depending upon the sample and solvents utilized.

The isolated, derivatized peptides retrieved are then analyzed using MS techniques. Both the quantity and sequence identity of the proteins from which the tagged peptides originated can be determined by automated multistage MS. This is achieved by the operation of the mass spectrometer in a dual mode in which it alternates in successive scans between measuring the relative quantities of peptides eluting from the capillary column and recording the sequence information of selected

peptides. Peptides are quantified by measuring in the MS mode the relative signal intensities for pairs of peptide ions of identical sequence that are tagged with the isotopically light or heavy forms of the compounds of the invention, respectively, and which therefore differ in mass by the mass differential encoded within the affinity-tagged reagent. Peptide sequence information is automatically generated by selecting peptide ions of a particular mass-to-charge ( $m/z$ ) ratio for collision-induced dissociation (CID) in the mass spectrometer operating in the  $MS^n$  mode. Using computer-searching algorithms, the resulting CID spectra are then automatically correlated with sequence databases to identify the protein from which the sequenced peptide originated. A combination of the results generated by MS and  $MS^n$  analyses of the differentially labeled peptide samples therefore determines the relative quantities as well as the sequence identities of the components of the protein mixtures in a single, automated operation. Alternatively, more accurate relative quantitation may be obtained by MS analysis of the isolated peptides with the mass spectrometer operating at MS mode only [see Automated LC/MS in Example 2: Instrumentation]

Apparatuses for performing MALDI-MS and techniques for using such apparatuses are described in International Publication No. WO 93/24835, US Patent 5,288,644, R. Beavis and B. Chait, *Proc. Natl. Acad. Sci. USA*, **87**:6873-6877 (1990); B. Chait and K. Standing, *Int. J. Mass Spectrom. Ion Phys.*, **40**:185 (1981) and Mamyrin et al, *Sov. Phys. JETP*, **37**:45 (1973), all of which are incorporated by reference herein. Briefly, the frequency tripled output of, e.g., a Q-switched Lumonics HY400 neodymium/yttrium aluminum garnet laser ("Nd-YAG") (355 nm, 10-nsec output pulse) is focused by a lens (12-inch focal length) through a fused silica window onto a sample inside the mass spectrometer. The product ions formed by the laser are accelerated by a static electric potential of 30 kV. The ions then drift down a 2-m tube maintained at a vacuum of 30  $\mu$ Pa and their arrival at the end of the tube is detected and recorded using, e.g., a Lecroy TR8828D transient recorder. The transient records of up to 200 individual laser shots are summed together and the resulting histogram is plotted as a mass spectrum. Peak centroid determinations and data reduction can be performed using a VAX workstation or other computer system. However, other

apparatuses and techniques are known and may be readily utilized for analysis of the peptides of the invention.

#### REAGENT KIT

5           The invention further provides a reagent kit for the analysis of proteins by mass spectral analysis. Typically, such a kit will contain one or more compounds of the invention. Most suitably, the kit will contain a set of substantially identical, differentially labeled (isotopically light and heavy) compounds. In one desirable embodiment, the kit will contain the compounds of the invention such that the  
10       polymer portion of the compound also serves as a solid support, e.g., a bead or resin. The kit may further contain one or more proteolytic enzymes, blocking reagents, solubilizing detergent cocktails, or wash solutions. Other suitable components will be readily apparent to one of skill in the art.

          The method and kit of the invention may be used for a variety of clinical and  
15       diagnostic assays, in which the presence, absence, deficiency or excess of a protein is associated with a normal or disease state. The method and kit of the invention can be used for qualitative and quantitative analysis of protein expression in cells and tissues. The method and kit can also be used to screen for proteins whose expression levels in cells or biological fluids are affected by a drug, toxin, environmental change, or by a  
20       change in condition or cell state, e.g., disease state, malignancy, site-directed mutation, gene therapy, or gene knockouts.

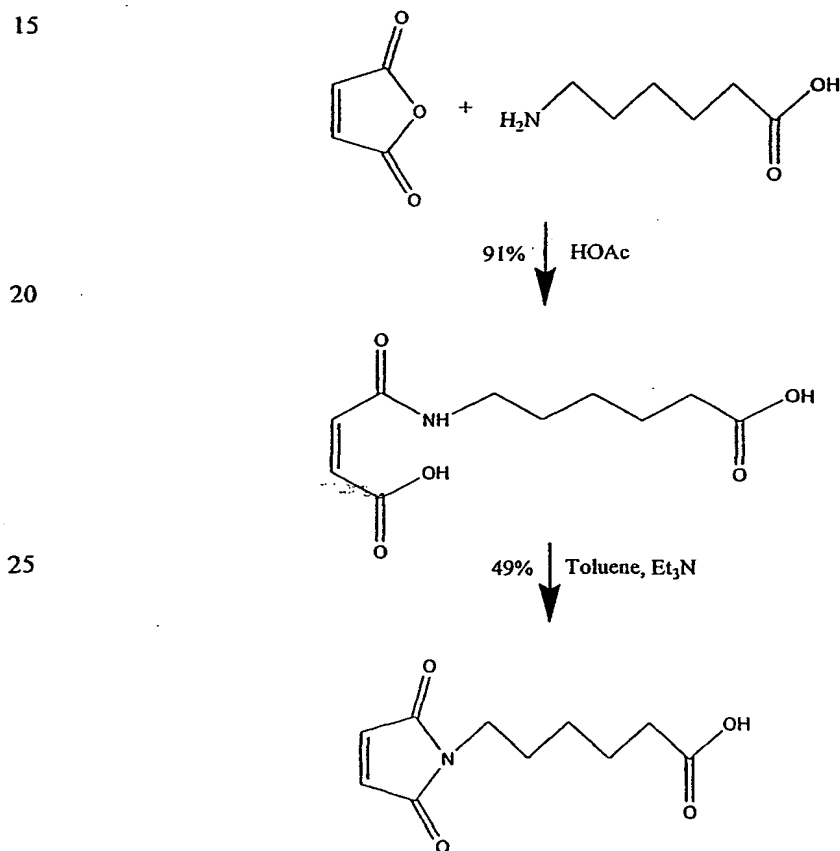
          The following examples are provided to illustrate the invention and do not limit the scope thereof. One skilled in the art will appreciate that although specific  
25       reagents and conditions are outlined in the following examples, modifications can be made which are meant to be encompassed by the spirit and scope of the invention.

## EXAMPLES

## EXAMPLE 1 - SYNTHESIS OF THE COMPOUND OF THE INVENTION

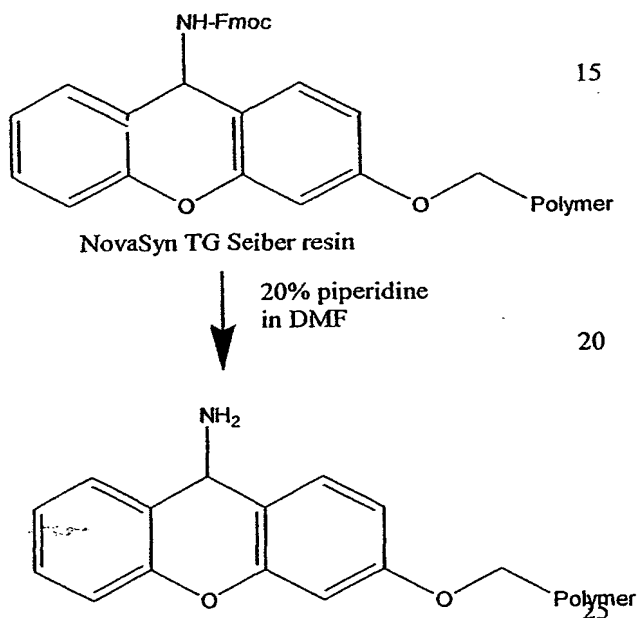
## A. Preparation Of Linker And Affinity Tag

A solution of maleic anhydride (0.98 g, 10.0 mmol in 15 ml of acetic acid) was added to a solution of 6-aminocaproic acid (1.31 g, 10 mmol in 5 ml of acetic acid). The resulting mixture was stirred at room temperature for two hours. After two hours, the mixture was heated to reflux (oil bath temperature about 110-120°C) for four and a half hours. The acetic acid was removed in vacuum and 3.3 g of a light yellow solid was obtained. This solid was chromatographed (20% ethyl acetate in hexanes, then 50% ethyl acetate in hexanes) and gave 0.92 g of pure target compound (6-(2,5-dioxo-2,5-dihydro-pyrrol-1-yl)-hexanoic acid; 43% yield). This reaction is illustrated in the scheme provided below, in which acetic acid is abbreviated as HOAc.



## B. Preparation Of Resin

The protected polymer, purchased commercially as NovaSyn TG Seiber resin (1 g, 0.15 mmol/g) was stirred in N, N-dimethylformamide (DMF) (8 mL) and then piperidine (2 mL) was added. The reaction mixture was stirred for ten minutes and then the solid was filtered and washed with methylene chloride and then dried under vacuum. This dry solid was then again stirred with piperidine (2 mL) in DMF (8 mL) for another ten minutes. The thin layer chromatography (TLC) was recorded and showed no trace of the fluorenylmethoxycarbonyl (Fmoc). The solid was then filtered and washed with methylene chloride, dried under low pressure to give about 1 g of the free amine polymer. This reaction is illustrated by the synthetic scheme below.



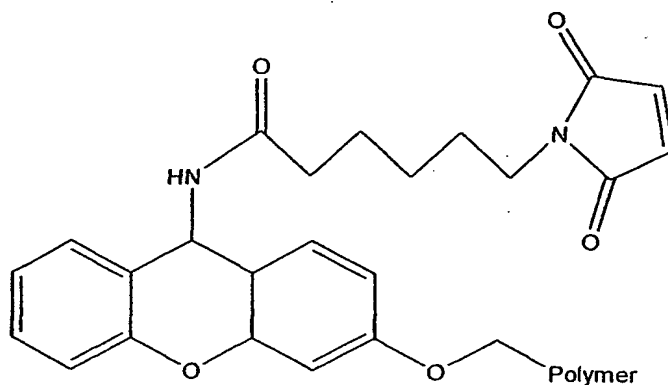
The polymer is a copolymer of polyethylene glycol and polystyrene.



## C. Preparation Of Compound Of The Invention

The deprotected polymer (1 g, 0.15 mmol/g) synthesized as described in part B was stirred in DMF (10 mL). To this mixture was added sequentially the compound which resulted from the reaction described in part A (0.095 g, 0.45 mmol),  
5 1-hydroxybenzotriazole (HOBT) (0.06 g, 0.45 mmol) and N, N-dicyclohexylcarbodiimide (DCC) (0.102 g, 0.5 mmol). The reaction mixture was stirred for three hours and the solid filtered and washed successively with ethyl acetate, ether and methylene chloride. The solid was then dried in vacuum and gave about 1 g of the product illustrated below (ALICE of the invention).

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## EXAMPLE 2 – INSTRUMENTATION

The present invention was carried out utilizing techniques and instrumentation known to those of skill in the art combined with a novel method of using the same. Specifically, data was obtained using automated LC/MS alone as well as using a novel automated 2-dimensional LC/LC/MS system using instrumentation available in the art. These instruments and methods of using the same are described below.

## A. Automated LC/MS

Automated LC/MS was accomplished using a LC/MS MicroMass Q-ToF<sup>2</sup> mass spectrometer (Micromass, Manchester, UK) equipped with an ABI 140 C microgradient syringe pump system (Applied Biosystems, Framingham, MA). The sample was injected onto a strong cation exchange (SCX) column, a 100 µm x 6 cm IntegraFrit column (New Objectives, Woburn, MA) packed with PolySULFOETHYL A, 12 µm, 300 Å (PolyLC Inc., Columbia, MD). The sample was then eluted onto a RP-C18 column, a 75 µm x 10 cm PicoFrit column (New Objectives, Woburn, MA) packed with YMC-Gel 10 µm C18 beads (YMC Inc., Wilmington, NC) using a solution of 500 mM KCl in 0.1 M acetic acid. The RP-C18 column was equilibrated with 96% acetic acid/4% ACN and then the following gradient was run: (i) 4-65% RP-B over 75 minutes, (ii) 65-98% RP-B over the next 7 minutes, (iii) a hold at 98% RP-B for 5 minutes, and (iv) 98-1% RP-B over the next 3 minutes at 250 µL/min. Mobile-phase buffers were for RP-A: 0.1 M acetic acid, 1 % ACN and RP-B: 0.1 M acetic acid, 90% ACN. Data was acquired in the MS mode only.

## B. Automated 2D-LC/MS/MS

Automated 2D-LC/MS/MS was accomplished using the system as shown in Figs 1A and 1B. Specifically, a 2D LC-MS/MS Finnigan LCQ Deca ion trap mass spectrometer was fitted with an Applied Biosystems 140C microgradient syringe pump system (Applied Biosystems, Framingham, MA), as the reverse phase pump (RP), and an Agilent 1100 series binary pump, as the strong cation exchange (SCX) and desalting pump. The pumps were attached to a VICI 10 port microbore two-position valve with a microelectric actuator (Valco Instruments CO Inc., Houston, TX). A strong cation exchange column, 50 x 1 mm PolySULFOETHYL A (PolyLC

Inc., Columbia, MD), was attached to port 9 and two 75 mm x 10 cm IntegraFrit columns (New Objectives, Woburn, MA) packed with YMC-Gel 10  $\mu$ m C18 beads (YMC Inc., Wilmington, NC) were attached between ports 2 and 5, and 7 and 10, respectively. Another 75  $\mu$ m x 3 cm C18 column packed in a PicoFrit column (New Objectives) was placed in between the titanium voltage union and the heated capillary of the mass spectrometer, to restore a loss of resolution from the valve and the titanium union.

Automation between the mass spectrometer, pumps and valve was accomplished using contact closures. First, the sample was loaded onto the SCX column using a Rheodyne injection valve (Rheodyne, Rohnert Park, CA) with the port valve at position 10 as shown in Fig. 1B so that any unbound peptides would bind to the RP-18 column and elute in fraction 0. With this dual C18 column design, while one RP-C18 column (column A in Fig. 1A) is being on-line with the mass spectrometer for peptide separation, the other C18 column (Column B in Fig. 1A) is being regenerated, loaded with peptide sample eluted from the SCX column and desalted. After each HPLC gradient run is completed, the positions of the two RP-C18 columns were switched over using the two-position ten-port valve (Fig. 1B) so that the time delay for equilibrating, sample loading from SCX and desalting was effectively eliminated. Peptide fractions were eluted from the SCX column onto one RP-C18 column using the following salt steps: (i) 5%, (ii) 10%, (iii) 15%, (iv) 20%, (v) 30%, (vi) 40%, (vii) 50%, (viii) 65%, (ix) 85%, (x) 98%, (xi) 98%, (xii) 98%, and (xiii) 98%, SCX-B:SCX-A, for 10 minutes at 1  $\mu$ L/min. Before each elution, 100% SCX-A was flowed at 1  $\mu$ L/min for 20 minutes to equilibrate the RP C18 column and after each salt elution, 100% SCX-A was flowed at 1  $\mu$ L/min for 20 minutes for elutions (i) to (iv), 25 minutes for elutions (v) and (vi), 30 minutes for elutions (vii) and (viii), and 35 minutes for elutions (ix) to (xiii). The flow was then slowed down to 200 nL/min for the remainder of time to rinse the salt from the RP C18 column. Peptides were eluted from one C18 column into the mass spectrometer using a linear RP gradient: a) 1-65% RP-B over 75 minutes, b) 65-98% RP-B over the next 7 minutes, c) a hold at 98% RP-B for 5 minutes, and d) 98-1% RP-B over the next 3

minutes at 400 nL/min. Mobile-phase buffers were, RP-A: 0.1 M acetic acid, 1 % ACN; RP-B: 0.1 M acetic acid, 90% ACN; SCX-A: 0.1 M acetic acid, 1% ACN; SCX-B: 500 mM KCl. (Figs. 1A and 1B).

### 5 EXAMPLE 3 - PREPARATION OF PROTEOMES FOR MS ANALYSIS

2 mg of bovine serum albumin (BSA) were solubilized in 200  $\mu$ L of 8 M urea, 200 mM ammonium bicarbonate, and 20 mM  $\text{CaCl}_2$ . 5  $\mu$ mole of tributyl phosphine (TBP) pre-dissolved in 20  $\mu$ L of acetonitrile (ACN) was added into the solubilized protein mixture and the resulting solution was incubated at 37°C for one hour. To the  
10 protein mixture was added 11  $\mu$ moles of MMTS and the mixture was vortexed for 10 minutes. The protein solution was diluted 1:1 with 100 mM ammonium bicarbonate and 40  $\mu$ g of Lys-C (2% w/w) were added. This mixture was then incubated at 37°C for 5 hours. The resulting solution was diluted 1:1 with water and then proteins were further digested with trypsin (2% w/w) at 37°C for 15 hours. The resulting peptide  
15 solution was dried and then reconstituted with 50% acetonitrile/200 mM sodium phosphate (pH 7.2). Disulfide bonds on the cysteine-containing peptides were reduced with TBP (5  $\mu$ moles) at 37°C for one hour. Then 50 mg of the ALICE resin (about 11.5  $\mu$ mole reactive sites) was added into the peptide solution and the solution vortexed for 1 hour at room temperature. The solutions were combined and loaded  
20 onto a column (glass type with teflon cockstop) and the resin was washed with the following solvent in sequence: 1) 5X 1 mL of 50% ACN, 2) 5X 1 mL of 30% ACN, 3) 5 X 1 mL of 90% ACN, 4) 5 X 1 mL of pure ACN, 5) 10 X 5 mL of dichloromethane (DCM).

Cysteine-containing peptides were then eluted from the resin with 5% TFA in  
25 DCM using continuous flow methodology. The resulting peptide solution was dried and reconstituted with 1% acetic acid in water. The reconstituted peptide solution was directly subjected to automated 2D-LC/MS/MS analysis (as described above) without further treatment. MS analysis combined with database searching yielded both identities and quantities of the proteins.

Samples were taken from the mixture before and after acid elution for MS analysis to compare the overall recovery of cysteine-containing peptides with or without using the ALICE approach. The results are provided below, with reference to the following published sequence of bovine serum albumin (using single letter amino acid code):

SEQ ID NO.1:

1	MKWVTFISLL	LLFSSATYSRG	VFRRDTHKSE	IAHRFKDLGE
41	EHFKGLVLIA	FSQYLQQCPF	DEHVKLVNEL	TEFAKTCVAD
81	ESHAGCEKSL	HTLFGDELCK	VASLRETYGD	MADCCEKQEP
121	ERNECFLSHK	DDSPDLPKLK	PDPNTLCDEF	KADEKKFWGK
161	YLYEIARRHP	YFYAPELLYY	ANKYNGVFQE	CCQAEDKGAC
201	LLPKIETMRE	KVLTSSARQR	LRCASIQKFG	ERALKAWSVA
241	RLSQKFPAE	FVEVTKLVTD	LTKVHKECCH	GDLLECADDR
281	ADLAKYICKN	QDTISSKLKE	CCDKPLLEKS	HCIAEVEKDA
321	IPENLPPLTA	DFAEDKDVCK	NYQEAKDAFL	GSFLYEYSRR
361	HPEYAVSVLL	RLAKEYEATL	EECCA KDDPH	ACYSTVFDKL
401	KHLVDEPQNL	IDQNCDQFEK	LGEYGFQNAL	IVRYTRKVPQ
441	VSTPTLVEVS	RSLGKVGTRC	CTKPESERMP	CTEDYLSLIL
481	NRLCVHEKT	PVSEKVTGCC	TESLVNRRPC	FSALTDETY
521	VPKAFDEKLF	TFHADICTLP	DTEKQIKKQT	ALVELLKHKP
561	KATEEQLKTV	MENFVAFVDK	CCAADDKEAC	FAVEGPKLVV
601	STQTALA			

Peptides identified from peptide mixtures before and after using ALICE for isolation  
of cysteine-containing peptides

Peptides identified by LC-MS/MS and database searching from the sample after enzymatic digestion but before reaction with ALICE (including both cysteine containing and non-cysteine containing peptides)		Peptides identified by LC-MS/MS and database searching from the final sample eluted from the ALICE resin (exclusively cysteine-containing peptides)	
Position, based on SEQ ID NO.1		Position, based on SEQ ID NO.1*	
508-523	76-88	508-523	460-468
402-412	437-451	89-100	<b>483-489</b>
106-117	89-100	267-280	123-130
198-204	298-309	106-117	286-297
310-318	267-280	581-587	499-507
161-167	375-386	45-65	199-204
123-130	499-507	310-318	198-204
286-297	360-371	76-88	300-309
460-468	562-568	588-597	387-399
421-433	123-138	52-65	375-386
529-544	95-100	139-151	319-340
300-309	588-597	413-420	<b>223-228</b>
413-420	533-544	529-544	469-482
598-607	548-557		
35-44	172-183		
45-65	319-340		
347-359	469-482		
341-353	435-451		
354-359	413-424		
168-180	387-399		
361-371	66-75		
581-597	549-557		
569-580	139-151		

- \* Two highlighted cysteine-containing peptides: CASIQK (residues 223-228) and
- 5 LCVLHEK (residues 483-489) were only detected from the final sample eluted from the ALICE resin.

This study demonstrated that nonspecific binding associated with the use of conventional reagents is not a problem using the compounds of the invention, since all the peptides eluted from the resin after washing are exclusively cysteine-containing peptides. This is because the compounds of the invention permit the use of much  
5 more stringent washing conditions, as compared to conventional ICAT reagents. Thus, the compounds of the invention provide lower "noise", better dynamic range and sensitivity in subsequent MS analysis.

More specifically, in this study, 33 out of 35 cysteines were captured. Only one Cys-containing peptide, YNGVFQECCQAEDK (residues 184 - 197 of SEQ ID  
10 NO.1) was not recovered either before or after isolation. CASIQK (residues 223-228 of SEQ ID NO.1) and LCVLHEK (residues 483-489 of SEQ ID NO.1) were only seen after isolation. This is likely due to the better dynamic range and sensitivity provided by the compound of the invention. Although not measured, overall recovery percentage is anticipated to be more than 75%. Steric hindrance in the capturing step  
15 is not a problem, since the peptides containing more than one cysteine were all uniformly modified by ALICE, the model compound of the invention. From all the CID experiments, no fragments observed were from the ALICE label, indicating that the compound would not interfere with the MS/MS experiments and subsequent protein identification by fragment-ion based database searching.

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#### EXAMPLE 4 - CAPTURING CYSTEINE-CONTAINING PEPTIDES USING ALICE, SIMPLE PROTEIN MIXTURES, AND AUTOMATED LC/MS and 2D-LC/MS

Two mixtures were prepared, each containing eight proteins. The following  
25 table illustrates the composition of these mixtures.

**Composition of two protein mixtures**

Protein Name	Protein Mixture A (nmol)	Protein Mixture B (nmol)
Lysozyme	10	50
$\alpha$ -lactalbumin	50	10
Ovalbumin	25	50
Catalase	50	25
$\beta$ -lactoglobulin	38	50
BSA	50	38
Ribonuclease	50	50
Trypsinogen	50	50

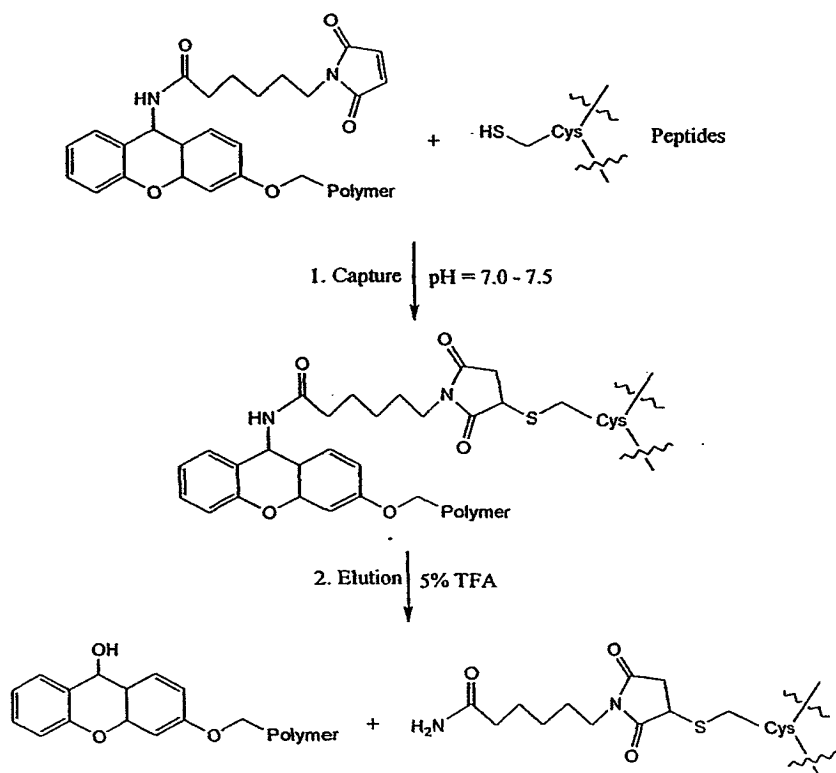
Protein mixture A and protein mixture B (323 nmol of total protein were solubilized, respectively, in 325  $\mu$ L of 6 M urea, 5% 3-[(3-cholamidopropyl)-  
5 dimethylammonio]-1-propanesulfonate (CHAPS), and 50 mM Tris HCl. 11.3  $\mu$ mole of tributyl phosphine (TBP) pre-dissolved in 6.3  $\mu$ L of isopropanol (IPA) was added to each solubilized protein mixture and the resulting solutions were incubated at 37°C for one hour. To each protein mixture was added 200  $\mu$ L of 50mM Tris-HCl (pH 8.0) and 34  $\mu$ mol of methanethiosulfonate (MTS) predissolved in 3.5  $\mu$ L of IPA, and  
10 the mixtures were reacted for 30 minutes. Each protein solution was diluted four times with 50 mM Tris-HCl (pH 8.0) and digested with trypsin (5% w/w) at 37°C for 16 hours. From the total peptide mixtures, 42% (21% from each mixture) was retained for future work, and the remaining 58% (187 nmol total protein) was dried and then reconstituted with 1.5 mL of 60% acetonitrile (ACN)/40% 100mM Tris-HCl  
15 (pH 7.0). Disulfide bonds on the cysteine-containing peptides were reduced by TBP (18.7  $\mu$ mol) at 37°C for one hour. Each solution was then vacuum concentrated for 10



minutes to remove excess TBP and ACN, and reconstituted to the previous volume using ACN. To each solution was added 55  $\mu$ mol of either light or heavy ALICE resins (3X TBP molar equivalent) and the solutions were stirred for 1 hour at room temperature. The reactions were quenched by the addition of  $\beta$ -mercaptoethanol (BME) to a final concentration of 1%.

The protein mixtures were then combined and loaded onto a column (fritted glass type with Teflon cockstop) and the resin was washed with the following solvent in sequence: (i) 50 mL of a 50:50 ACN:water solution, (ii) 50 mL of pure ACN, (iii) 50 mL of a 50:50 ACN:dichloromethane (DCM) solution, and (iv) 50 mL of pure DCM.

Cysteine-containing peptides were isolated by elution with 3 x 5 mL of 5% TFA in DCM using continuous flow methodology, 15 minute incubations with intermittent shaking, then 15 mL of continuous flow. The resulting peptide solution was dried and reconstituted with 2% ACN in 1% acetic acid/water. The reconstituted peptide solution was directly subjected to HPLC-MS MicroMass Q-ToF<sup>2</sup> instrument (MicroMass, Manchester, UK) and 2D-LC-MS/MS (Finnigan LCQ Deca, Finnigan Corporation, San Jose, CA) analysis without further treatment. These analyses, combined with database searching, yielded both identities and quantities of the proteins. The chemical reactions for the isolation of cysteine-containing peptides are illustrated in the following scheme.



- 5 The results of the mass-spectrometric analysis are provided in the following table. In this table, M# = oxidized methionine residue; C\* = light and heavy ALICE labeled cysteine residue.

Table - Sequence identification and quantitation of the components of a protein mixture using ALICE.

Protein Name	Peptide Mass// Charge State	Peptide Sequence identified/ SEQ ID NO:	Obs'd Ratio/ Mean $\pm$ SD	Exp. Ratio	% Error
$\alpha$ -lactoalbumin	432.20// 2	(K)C*EVFR(E) SEQ ID NO:3	4.97	5	0.6
$\beta$ -lactoglobulin/	1107.84// 3	(K)YLLFC*M#ENSAEPEQSLVC*QC *LVR(T) : SEQ ID NO:4	0.76 0.76 $\pm$ 0.01	0.76	0.3
	934.94// 2	(R)LSFNPTQLEEQC*HI(-) : SEQ ID NO:5	0.77		
Catalase	654.34// 2	(R)LC*ENIAGHLK(D) : SEQ ID NO: 6	2.1 0.02 $\pm$ 0.09	2	1
	436.56// 3	(R)LC*ENIAGHLK(D): SEQ ID NO:6	1.93		
	979.00// 2	(R)LGPNYLQIPVNC*PYR(A) : SEQ ID NO:7	2.01		
Lysozyme	1062.49// 1	(R)C*ELAAAM#K(R) :SEQ ID NO:8	0.2	0.2	0.2
Ovalbumin	739.80// 2	(A)SM#EFCFDVFK(E) :SEQ ID NO:9	0.61 0.58 $\pm$ 0.05	0.5	16
	700.85// 2	(R)ADHPFLFC*IK(H): SEQ ID NO: 10	0.6		
	467.57// 3	(R)ADHPFLFC*IK(H) :SEQ ID NO:10	0.52		
	838.44// 2	(R)YPILPEYLQC*VK(E) SEQ ID NO:11	0.59		
Ribonuclease	1189.08// 2	(K)HIIVAC*EGNPYVPVHFDASV(-) SEQ ID NO:12	1.08 1.00 $\pm$ 0.11	1	0.4
	793.06// 3	(K)HIIVAC*EGNPYVPVHFDASV(-) SEQ ID NO:12	1.16		

Table (cont'd)

Protein Name	Peptide Mass// Charge State	Peptide Sequence identified/ SEQ ID NO:	Obs'd Ratio/ Mean $\pm$ SD	Exp. Ratio	% Error
	595.04// 4	(K)HIIIVAC*EGNPYVPVHFDASV(-) SEQ ID NO:12	1.17		
	706.60// 4	(R)C*KPVNTFVHESLADVQAVC*S QK(N) SEQ ID NO:13	0.89		
	922.40// 2	(A)CEGNPYVPVHFDASV(-) aa 6-22 of SEQ ID NO:12	1.03		
	608.63// 3	(F)VHESLADVQAVCSQK(N) aa 6-24 of SEQ ID NO:12	0.96		
	865.5//1	(K)HIIIVAC*(E) aa 1-8 of SEQ ID NO:14	1.03		
	433.25// 2	(K)HIIIVAC*(E) aa 1-8 of SEQ ID NO:14	0.9		
	1239.5// 1	(Y)STM#SITDC*R(E) SEQ ID NO:14	0.9		
	620.25// 2	(Y)STM#SITDC*R(E) SEQ ID NO:14	0.84		
Trypsinogen	580.3//2	(A)PILSDSSC*K(S) aa 5-15 of SEQ ID NO:15	0.87 1.02 $\pm$ 0.10	1	2
	1230.61// 1	(K)APILSDSSC*K(S) aa 4-15 of SEQ ID NO:15	1.01		
	615.80// 2	(K)APILSDSSC*K(S) aa 4-15 of SEQ ID NO:15	1.18		
	892.95// 2	(K)C*LKAPILSDSSC*K(S) SEQ ID NO:15	1.02		
	595.63// 3	(K)C*LKAPILSDSSC*K(S) SEQ ID NO:15	1.04		
	958.41// 2	(K)DSC*QGDSGGPVVC*SGK(L) SEQ ID NO:16	0.98		

Table (cont'd)

Protein Name	Peptide Mass// Charge State	Peptide Sequence identified/ SEQ ID NO:	Obs'd Ratio/ Mean $\pm$ SD	Exp. Ratio	% Error
BSA	1141.6// 1	(C)C*TESLVNR(R) aa 497-506 of SEQ ID NO:1	1.5 1.35 $\pm$ 0.10	1.32	2.3
	566.25// 2	(C)C*TESLVNR(R) aa 497-506 of SEQ ID NO:1	1.28		
	623.35// 2	(H)TLFGDELK*(V) aa 92-102 of SEQ ID NO:1	1.21		
	1194.02// 12	(K)C*C*AADDKEAC*FAVEGPK(L) aa 577-595 of SEQ ID NO:1	1.24		
	796.35// 3	(K)C*C*AADDKEAC*FAVEGPK(L) aa 577-595 of SEQ ID NO:1	1.23		
	722.83// 2	(K)C*C*TESLVNR(R) aa 496-506 of SEQ ID NO:1	1.34		
	650.30// 3	(K)DDPHAC*YSTVFDK(L)(H) aa 386-402 of SEQ ID NO:1	1.35		
	630.80// 2	(K)EAC*FAVEGPK(L) aa 584-595 of SEQ ID NO:1	1.3		
	533.25// 3	(K)EC*C*DKPLLEK(S) aa 300-311 of SEQ ID NO:1	1.41		
	911.50// 1	(K)GAC*LLPK(I) aa 198-206 of SEQ ID NO:1	1.48		
	638.80// 2	(K)LFTFHADIC*(T) aa 525-535 of SEQ ID NO:1	1.35		
	638.80// 2	(K)LFTFHADIC*(T) aa 525-535 of SEQ ID NO:1	1.51		
	613.65// 3	(K)LKEC*C*DKPLLEK(S) aa 298-311 of SEQ ID NO:1	1.51		
	577.28// 3	(K)LKPDPNTLC*DEFK(A) aa 139-153 of SEQ ID NO:1	1.21		
	786.89// 2	(K)SLHTLFGDELK*(V) aa 89-102 of SEQ ID NO:1	1.35		

Table (cont'd)

Protein Name	Peptide Mass// Charge State	Peptide Sequence identified/ SEQ ID NO:	Obs'd Ratio/ Mean $\pm$ SD	Exp. Ratio	% Error
	524.92// 3	(K)SLHTLFGDELC*K(V) aa 89-102 of SEQ ID NO:1	1.35		
	885.37// 2	(K)TC*VADESHAGC*EK(S) aa 76-90 of SEQ ID NO:1	1.52		
	590.58// 3	(K)TC*VADESHAGC*EK(S) aa 76-90 of SEQ ID NO:1	1.52		
	591.62// 3	(K)VTKC*C*TESLVNR(R) aa 493-506 of SEQ ID NO:1	1.19		
	798.86// 2	(K)YIC*DNQDTISSK(L) aa 286-299 of SEQ ID NO:1	1.36		
	1027.43 //2	(K)YNGVFQEC*C*QAEDK(G) aa 184-199 of SEQ ID NO:1	1.2		
	859.43// 1	(R)C*ASIQK(F) aa 223-230 of SEQ ID NO:1	1.46		
	430.21// 2	(R)C*ASIQK(F) aa 223-230 of SEQ ID NO:1	1.3		
	1051.56 //1	(R)LC*VLHEK(T) aa 481-488 of SEQ ID NO:1	1.35		
	526.284 //2	(R)LC*VLHEK(T) aa 481-488 of SEQ ID NO:1	1.27		
	947.45// 2	(R)M#PC*TEDYLSLILNR(L) aa 468-482 of SEQ ID NO:1	1.36		
	631.97// 3	(R)M#PC*TEDYLSLILNR(L) aa 468-482 of SEQ ID NO:1	1.25		
	1027.97 //2	(R)NEC*FLSHKDDSPDLPK(L) aa 123-140 of SEQ ID NO:1	1.27		
	1017.50 //2	(R)RPC*FSALTPDETYVPK(A) aa 505-521 of SEQ ID NO:1	1.41		
	678.672 //3	(R)RPC*FSALTPDETYVPK(A) aa 505-521 of SEQ ID NO:1	1.39		

This study demonstrated that quantification by ALICE is accurate after taking into account the following factors: isotopic impurity of the heavy ALICE; different elution profile of the same peptides modified by heavy and light ALICE; non-specific enzymatic cleavage. This improved quantitation accuracy by ALICE is even more  
5 evident when multiple cysteine-containing peptides are present. Peptides without any cysteine residue were rarely seen in the final captured peptide mixture since more stringent washing conditions completely removed non-specifically bound species. Furthermore, the use of large amounts of organic solvents also minimized the loss of peptides throughout the procedure. Finally, simplification of the peptide mixture by  
10 isolating cysteine-containing peptides in combination with the novel automated 2D-LC/MS design increase the overall sample loading capacity, the speed of sample analysis and the dynamic range and sensitivity of the MS analysis of protein mixtures. This experiment also further confirmed that reaction between ALICE and cysteine-containing peptides is efficient and stoichiometric and the effect of steric hindrance is  
15 not a concern since peptides with more than one cysteine residue were modified completely by ALICE. For example, a tryptic peptide with three cysteine residues derived from lysozyme (NLC\*NIPC\*SALLSSDITASVNC\*AK, SEQ ID NO:2) was uniformly labeled with either heavy or light ALICE (the mass difference (not shown) between this heavy and light mass pairs is exactly 30 Da). Both light and heavy  
20 ALICE labeled peptides were effectively picked by the automated 2D-LC/LC/MS system for MS/MS analysis even though the peak intensity for the light ALICE labeled peptide is very low. Subsequent database searching identified the peptide as NLC\*NIPC\*SALLSSDITASVNC\*AK [SEQ ID NO:2] with cysteine residues modified by light and heavy ALICE, respectively.

25

All publications cited in this specification are incorporated herein by reference herein. While the invention has been described with reference to a particularly preferred embodiment, it will be appreciated that modifications can be made without departing from the spirit of the invention. Such modifications are intended to fall  
30 within the scope of the appended claims.

## WHAT IS CLAIMED IS:

1. A method for the analysis of mixtures containing proteins, said method comprising the steps of:

- (a) reducing the disulfide bonds in the proteins of a sample, thereby providing thiol groups in cysteine-containing proteins;
- (b) blocking free thiols with a blocking reagent in the sample;
- (c) digesting the proteins in the sample to provide peptides;
- (d) reducing the disulfide bonds in the digested peptides, thereby providing thiol groups in cysteine-containing peptides for reaction;
- (e) reacting cysteine-containing peptides in the sample with a reagent, wherein said reagent comprises a thiol-specific reactive group which is attached to a polymer tag via a linker, wherein the linker can be differentially labeled with stable isotopes and wherein the polymer tag forms a covalent bond with the cysteine-containing peptides;
- (f) washing the polymer-bound peptides to remove non-covalently bound species;
- (g) eluting the cysteine-containing peptides; and
- (h) subjecting the eluted peptides to quantitative mass spectrometry (MS) analysis.

2. The method according to claim 1, wherein said method further comprises the steps of:

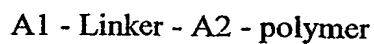
- performing steps (a) to (d) on a second sample;
- reacting cysteine-containing labels in the second sample with a stable isotope-labeled form of the reagent, wherein in reacting step (e), the reagent used is a non-isotope labeled form the reagent;
- mixing the peptides of the reacted sample following step (e) and the reacted second sample; and
- performing steps (g) and (h) on the peptides in the mixture.



3. The method according to claim 1, wherein the reagent comprises a thiol-specific reactive group is selected from the group consisting of  $\alpha$ -haloacetyl and maleimide.

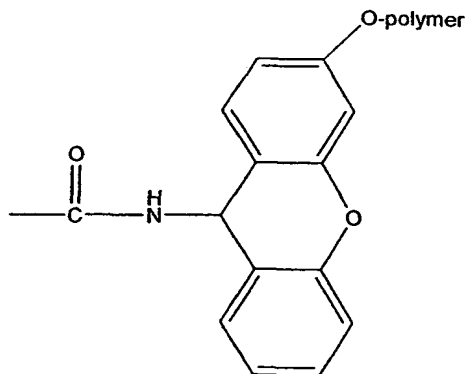
4. The method according to claim 1, wherein the blocking reagent is methyl methane thiosulfonate.

5. The method according to claim 1, wherein the reagent has the formula:



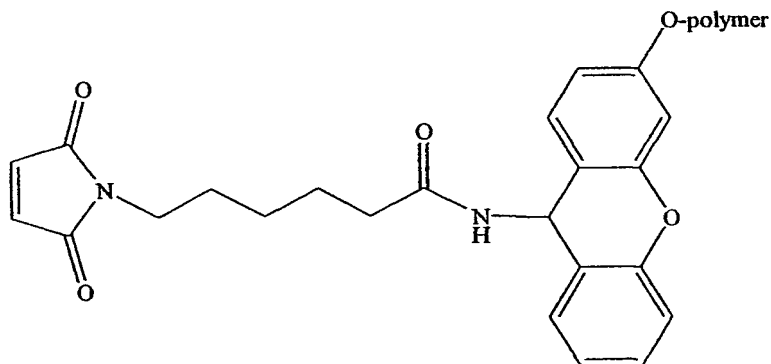
wherein A1 is the thiol-reactive group and A2 is an acid labile group to which the polymer is bound.

6. The method according to claim 5, wherein the acid-labile group bound to the polymer has the structure:

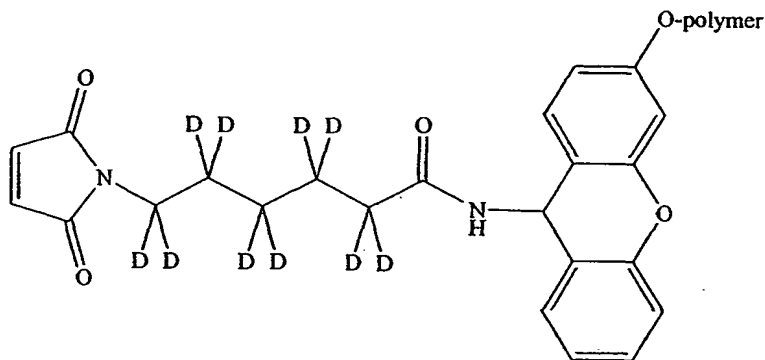


7. The method according to claim 5, wherein the polymer in the reagent is a polymer resin.

8. The method according to claim 7, wherein the polymer resin is a homopolymer or heteropolymer comprising a polymer selected from the group consisting of polystyrene and polyethylene glycol.
9. The method according to claim 8, wherein the linker contains a substitution of at least six hydrogen atoms with a stable isotope.
10. The method according to claim 9, wherein the linker contains ten stable isotopes.
11. The method according to claim 9, wherein the stable isotope is deuterium.
12. The method according to claim 1, wherein the non-isotope labeled reagent is

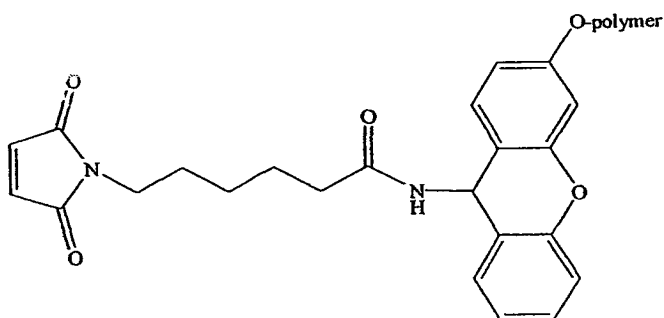


13. The method according to claim 1, wherein the isotope labeled reagent has the formula:

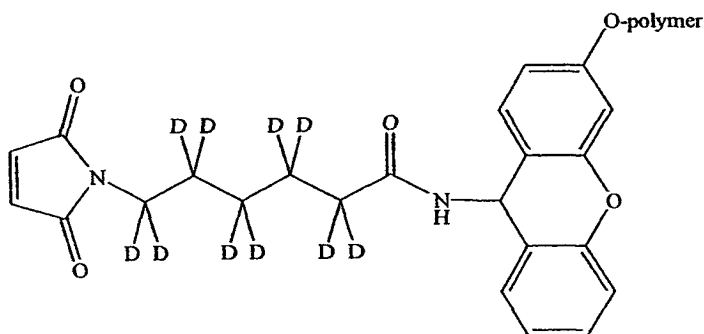


14. The method according to claim 1, wherein the eluted peptides are subjected to high-performance liquid chromatography-mass spectrometry (MS) analysis, two-dimensional liquid chromatography MS, or MS/MS analysis.
15. The method according to claim 1, wherein the proteins are digested using trypsin.
16. A compound useful for capturing cysteine-containing peptides, which is selected from the group consisting of a thiol-specific reactive group attached to a non-biological polymer via a linker.
17. The compound according to claim 16, wherein the linker contains a substitution of at least six atoms with a stable isotope.
18. The compound according to claim 16, wherein the linker contains ten stable isotopes.
19. The compound according to claim 17, wherein the stable isotope is deuterium.

20. The compound according to claim 16, selected from the group consisting of:



and



21. A reagent kit for the analysis of proteins by mass spectral analysis that comprises a compound of claim 16.

22. The reagent kit of claim 21 which comprises a set of substantially identical differentially labeled cysteine-tagging reagents.

23. The reagent kit of claim 22 further comprising one or more proteolytic enzymes for use in digestion of proteins to be analyzed.

Fig. 1B

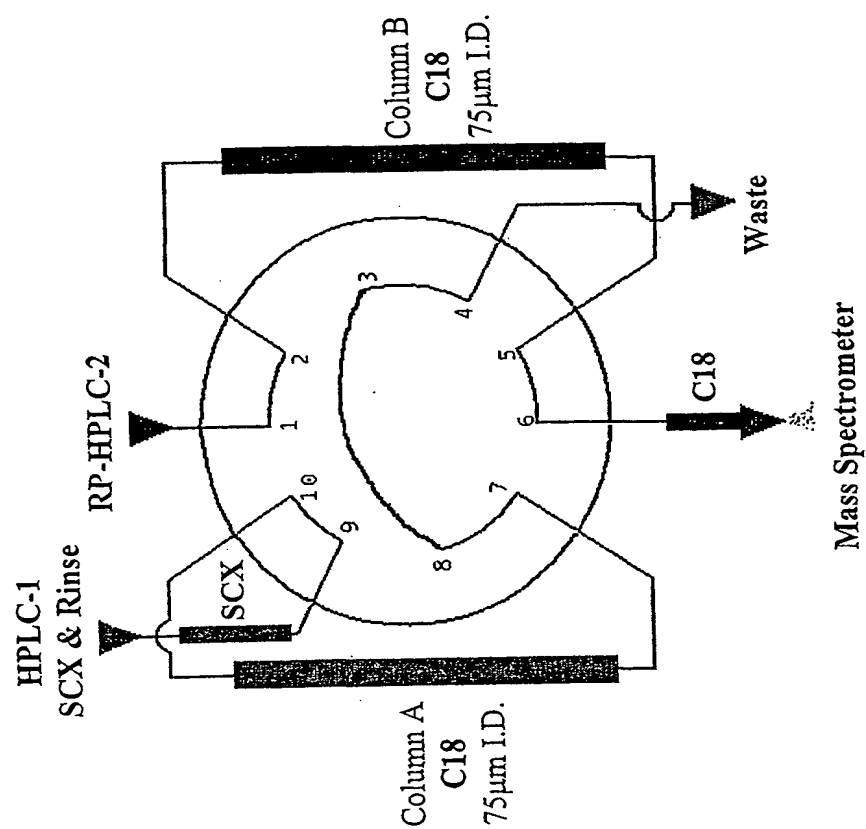
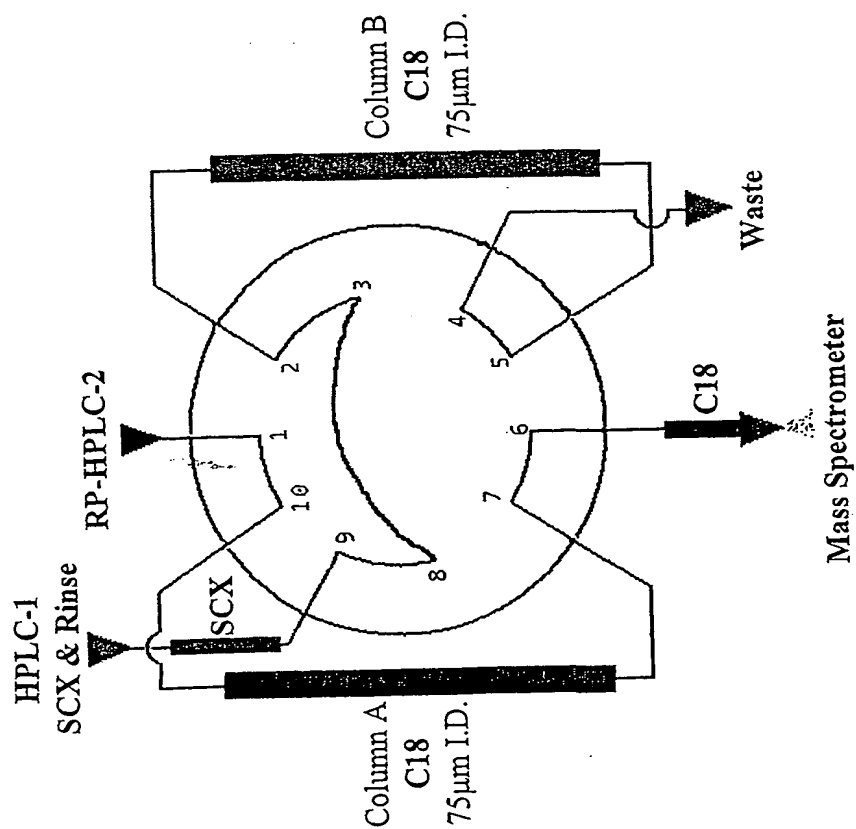


Fig. 1A



## SEQUENCE LISTING

<110> Genetics Institute, Inc.

<120> ACID-LABILE ISOTOPE-CODED EXTRACTANT (ALICE) AND ITS USE IN  
QUANTITATIVE MASS SPECTROMETRIC ANALYSIS OF PROTEIN MIXTURES

<130> GI5412APCT

<150> 60/242,643

<151> 2000-10-23

<160> 16

<170> PatentIn version 3.1

<210> 1

<211> 604

<212> PRT

<213> Bovine Serum Albumin

<400> 1

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Ala His Arg Phe Lys Asp Leu Gly Glu Glu His Phe Lys Gly Leu Val  
35 40 45

Leu Ile Ala Phe Ser Gln Tyr Leu Gln Gln Cys Pro Phe Asp Glu His  
50 55 60

Val Lys Leu Val Asn Glu Leu Thr Glu Phe Ala Lys Thr Cys Val Ala  
65 70 75 80

Asp Glu Ser His Ala Gly Cys Glu Lys Ser Leu His Thr Leu Phe Gly  
85 90 95

Asp Glu Leu Cys Lys Val Ala Ser Leu Arg Glu Thr Tyr Gly Asp Met  
100 105 110

Ala Asp Cys Cys Glu Lys Gln Glu Pro Glu Arg Asn Glu Cys Phe Leu  
115 120 125

Ser His Lys Asp Asp Ser Pro Asp Leu Pro Lys Leu Lys Pro Asp Pro  
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Asn Thr Leu Cys Asp Glu Phe Lys Ala Asp Glu Lys Lys Phe Trp Gly

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	165	170	175
Glu Leu Leu Tyr	Tyr Ala Asn Lys Tyr	Asn Gly Val Phe	Gln Glu Cys
	180	185	190
Cys Gln Ala Glu	Asp Lys Gly Ala Cys	Leu Leu Pro Lys	Ile Glu Thr
	195	200	205
Met Arg Glu Lys	Val Leu Thr Ser Ser	Ala Arg Gln Arg	Leu Arg Cys
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Ala Ser Ile Gln	Lys Phe Gly Glu Arg	Ala Leu Lys Ala	Trp Ser Val
	225	230	235
Ala Arg Leu Ser	Gln Lys Phe Pro Lys	Ala Glu Phe Val	Glu Val Thr
	245	250	255
Lys Leu Val Thr	Asp Leu Thr Lys	Val His Lys Glu	Cys Cys His Gly
	260	265	270
Asp Leu Leu Glu	Cys Ala Asp Asp	Arg Ala Asp Leu	Ala Lys Tyr Ile
	275	280	285
Cys Lys Asn Gln	Asp Thr Ile Ser Ser	Lys Leu Lys Glu	Cys Cys Asp
	290	295	300
Lys Pro Leu Leu	Glu Lys Ser His Cys	Ile Ala Glu Val	Glu Lys Asp
	305	310	315
Ala Ile Pro Glu	Asn Leu Pro Pro	Leu Thr Ala Asp	Phe Ala Glu Asp
	325	330	335
Lys Val Cys Lys	Asn Tyr Gln Glu	Ala Lys Asp Ala	Phe Leu Gly Ser
	340	345	350
Phe Leu Tyr Glu	Tyr Ser Arg Arg	His Pro Glu Tyr	Ala Val Ser Val
	355	360	365
Leu Leu Arg Leu	Ala Lys Glu Tyr	Glu Ala Thr Leu	Glu Glu Cys Cys
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Ala Lys Asp Asp	Pro His Ala Cys	Tyr Ser Thr Val	Phe Asp Lys Leu
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Lys His Leu Val	Asp Glu Pro Gln	Asn Leu Ile Asp	Gln Asn Cys Asp
	405	410	415
Gln Phe Glu Lys	Leu Gly Glu Tyr	Gly Phe Gln Asn	Ala Leu Ile Val
	420	425	430

Arg Tyr Thr Arg Lys Val Pro Gln Val Ser Thr Pro Thr Leu Val Glu  
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 Val Ser Arg Ser Leu Gly Lys Val Gly Thr Arg Cys Cys Thr Gly Pro  
 450 455 460  
 Glu Ser Glu Arg Met Pro Cys Thr Glu Asp Tyr Leu Ser Ile Leu Asn  
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 485 490 495  
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 500 505 510  
 Thr Asp Glu Thr Tyr Val Pro Lys Ala Phe Asp Glu Lys Leu Phe Thr  
 515 520 525  
 Phe His Ala Asp Ile Cys Thr Leu Pro Asp Thr Glu Lys Gln Ile Lys  
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 Lys Gln Thr Ala Leu Val Glu Leu Leu Lys His Lys Pro Lys Ala Thr  
 545 550 555 560  
 Glu Glu Gln Leu Lys Thr Val Met Glu Asn Phe Val Ala Phe Val Asp  
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<210> 3  
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<213> Peptide from alpha-lactoalbumin

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<210> 4

<211> 25

<212> PRT

<213> Peptide from beta-lactoglobulin

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<211> 15

<212> PRT

<213> Peptide from beta-lactoglobulin

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<210> 15  
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<400> 15

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<210> 16  
<211> 18  
<212> PRT  
<213> Peptide from trypsinogen

<400> 16

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Lys Leu

(12) INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

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(71) Applicant: GENETICS INSTITUTE, LLC. [US/US];  
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(81) Designated States (*national*): AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, PH, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, TZ, UA, UG, UZ, VN, YU, ZA, ZW.

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(72) Inventors: QIU, Yongchang; 10 Old Colony Lane #7, Arlington, MA 02476 (US). WANG, Jack, H.; 7 Hathaway Road, Lexington, MA 02420 (US). HEWICK, Rodney, M.; 42 Commonwealth Avenue #4, Boston, MA 02116 (US).

(74) Agents: KODROFF, Cathy, A. et al.; Howson and Howson, Spring House Corporate Center, P.O. Box 457, Spring House, PA 19477 (US).

For two-letter codes and other abbreviations, refer to the "Guidance Notes on Codes and Abbreviations" appearing at the beginning of each regular issue of the PCT Gazette.

(54) Title: ACID-LABILE ISOTOPE-CODED EXTRACTANT (ALICE) IN MASS SPECTROMETRIC ANALYSIS

(57) Abstract: The method of the invention provides novel compounds, termed acid-labile isotope-coded extractants (ALICE), for quantitative mass spectrometric analysis of protein mixtures. The compounds contain a thiol-reactive group that is used to capture cysteine-containing peptides from all peptide mixtures, an acid-labile linker, and a non-biological polymer. One of the two acid-labile linkers is isotopically labeled and therefore enables the direct quantitation of peptides/proteins through mass spectrometric analysis. Because no functional proteins are required to capture peptides, a higher percentage of organic solvent can be used to solubilize the peptides, particularly hydrophobic peptides, through the binding, washing and eluting steps, thus permitting much better recovery of peptides. Moreover, since the peptides are covalently linked to the non-biological polymer (ALICE), more stringent washing is allowed in order to completely remove non-specifically bound species. Finally, peptides captured by ALICE are readily eluted from the polymer support under mild acid condition with high yield and permit the direct down stream mass spectrometric analysis without any further sample manipulation. In combination with our novel dual column two dimensional liquid chromatography-mass spectrometry (2D-LC-MS/MS) design, the ALICE procedure proves to a general approach for quantitative mass spectrometric analysis of protein mixtures with better dynamic range and sensitivity.

WO 02/048717 A3

## INTERNATIONAL SEARCH REPORT

 Internat Application No  
 PCT/US 01/50745

 A. CLASSIFICATION OF SUBJECT MATTER  
 IPC 7 G01N33/68 G01N33/58 G01N33/532 C07D405/12

According to International Patent Classification (IPC) or to both national classification and IPC

## B. FIELDS SEARCHED

 Minimum documentation searched (classification system followed by classification symbols)  
 IPC 7 G01N

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

EPO-Internal, WPI Data, PAJ, BIOSIS, MEDLINE, EMBASE, CHEM ABS Data, SCISEARCH

## C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	US 5 306 562 A (REGENTS OF THE UNIVERSITY OF MINNESOTA) 26 April 1994 (1994-04-26) the whole document	16-23
A	KALEF E ET AL: "Arsenical-based affinity chromatography of vicinal dithiol-containing proteins: purification of L1210 leukemia cytoplasmic proteins and the recombinant rat c-erb A(beta)1 T3 receptor" ANALYTICAL BIOCHEMISTRY, ACADEMIC PRESS, SAN DIEGO, CA, US, vol. 212, no. 2, 1 August 1993 (1993-08-01), pages 325-334, XP002122033 ISSN: 0003-2697 the whole document	1-15

☒ Further documents are listed in the continuation of box C.

☒ Patent family members are listed in annex.

## \* Special categories of cited documents:

- \*A\* document defining the general state of the art which is not considered to be of particular relevance
- \*E\* earlier document but published on or after the international filing date
- \*L\* document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)
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- \*Y\* document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.
- \* & \* document member of the same patent family

Date of the actual completion of the international search

21 November 2002

Date of mailing of the international search report

28/11/2002

Name and mailing address of the ISA

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## INTERNATIONAL SEARCH REPORT

Intern al Application No

PCT/US 01/50745

## C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	SECHI S & CHAIT BT: "Modification of Cysteine Residue by Alkylation. A Tool in Peptide Mapping and Protein Identification" ANALYTICAL CHEMISTRY, AMERICAN CHEMICAL SOCIETY. COLUMBUS, US, vol. 70, no. 24, 15 December 1998 (1998-12-15), pages 5150-5158, XP002192468 ISSN: 0003-2700 the whole document	1-15
P,A	EP 1 059 531 A (NORDHEIM A., CAHILL M.) 13 December 2000 (2000-12-13) claims 1-16	1-15

# INTERNATIONAL SEARCH REPORT

In **International application No.**  
**PCT/US 01/50745**

## Box I Observations where certain claims were found unsearchable (Continuation of item 1 of first sheet)

This International Search Report has not been established in respect of certain claims under Article 17(2)(a) for the following reasons:

1. ☐ Claims Nos.:  
because they relate to subject matter not required to be searched by this Authority, namely:
2. ☒ Claims Nos.:  
because they relate to parts of the International Application that do not comply with the prescribed requirements to such an extent that no meaningful International Search can be carried out, specifically:  
see FURTHER INFORMATION sheet PCT/ISA/210
3. ☐ Claims Nos.:  
because they are dependent claims and are not drafted in accordance with the second and third sentences of Rule 6.4(a).

## Box II Observations where unity of invention is lacking (Continuation of item 2 of first sheet)

This International Searching Authority found multiple inventions in this international application, as follows:

1. ☐ As all required additional search fees were timely paid by the applicant, this International Search Report covers all searchable claims.
2. ☐ As all searchable claims could be searched without effort justifying an additional fee, this Authority did not invite payment of any additional fee.
3. ☐ As only some of the required additional search fees were timely paid by the applicant, this International Search Report covers only those claims for which fees were paid, specifically claims Nos.:
4. ☐ No required additional search fees were timely paid by the applicant. Consequently, this International Search Report is restricted to the invention first mentioned in the claims; it is covered by claims Nos.:

### Remark on Protest

- ☐ The additional search fees were accompanied by the applicant's protest.
- ☐ No protest accompanied the payment of additional search fees.



## FURTHER INFORMATION CONTINUED FROM PCT/ISA/ 210

## Continuation of Box I.2

Present claims 16-19 relate to an extremely large number of possible compounds.

Support within the meaning of Article 6 PCT and/or disclosure within the meaning of Article 5 PCT is to be found, however, for only a very small proportion of the compounds claimed. In the present case, the claims so lack support, and the application so lacks disclosure, that a meaningful search over the whole of the claimed scope is impossible.

Consequently, the search has been carried out for those parts of the claims which appear to be supported and disclosed, namely those parts relating to the compounds disclosed in claim 20.

The applicant's attention is drawn to the fact that claims, or parts of claims, relating to inventions in respect of which no international search report has been established need not be the subject of an international preliminary examination (Rule 66.1(e) PCT). The applicant is advised that the EPO policy when acting as an International Preliminary Examining Authority is normally not to carry out a preliminary examination on matter which has not been searched. This is the case irrespective of whether or not the claims are amended following receipt of the search report or during any Chapter II procedure.

# INTERNATIONAL SEARCH REPORT

International Application No

PCT/US 01/50745

Patent document cited in search report		Publication date	Patent family member(s)	Publication date
US 5306562	A	26-04-1994	US 5117009 A EP 0473411 A1 JP 4288070 A	26-05-1992 04-03-1992 13-10-1992
EP 1059531	A	13-12-2000	EP 1059531 A1 AU 4927000 A WO 0077520 A1	13-12-2000 02-01-2001 21-12-2000